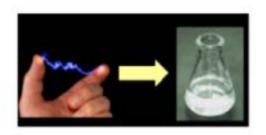
"PRACTICAL MARINE APPLICATIONS OF RENEWABLE METHANOL"





Paul Wuebben
Senior Director, Renewable Fuels
CARBON RECYCLING INTERNATIONAL

Symposium on Evolution of Maritime Fuels, Ship Speed and Operational Efficiency

USC Sea Grant Program

Loker HC Research Institute, July 21, 2014



OUTLINE

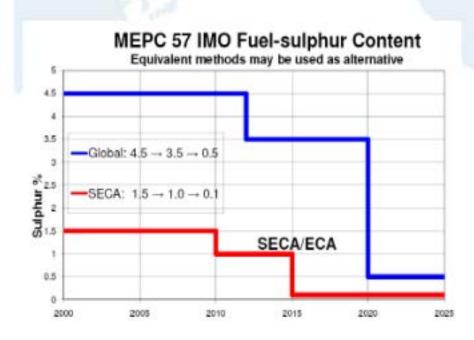
- ENERGY AND REGULATORY CONTEXT
- SUSTAINABLE METHANOL
 - ► CRI'S CO₂ TO METHANOL PLANT
- METHANOL SHIP DEMONSTRATIONS
- METHANOL OUTLOOK
- CONCLUSIONS

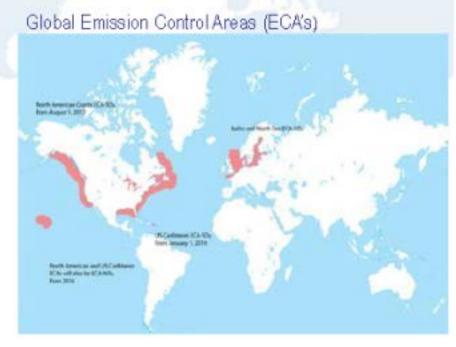
BEWARE OF ENERGY COMPLACENCY:

- For 70% of the world's oil fields, according to the IEA, the rates of decline of these fields is now observed at 6.2%/yr.
- Since the year 2000, global upstream CAPEX grew 3 x from \$250 B to \$700 while total supply grew by only 12%.
- While No. America now accounts for 20% of the world's oil production, it accounts for 50% of the world's CAPEX for oil.
- 2013 global energy demand reached another all time peak
- Even if the US becomes "energy self-sufficient" for the next decade, it will remain coupled to the global price of crude oil.
- Exxon Mobil has removed staff from two large fields in the south of Iraq.
- Russia: oil and NG wildcard / supply insecurities
- Middle East volatility is escalating (Syria, Iran, Iraq, etc.)
- Unabated global demographics

Marine Fuel Innovation Driver:

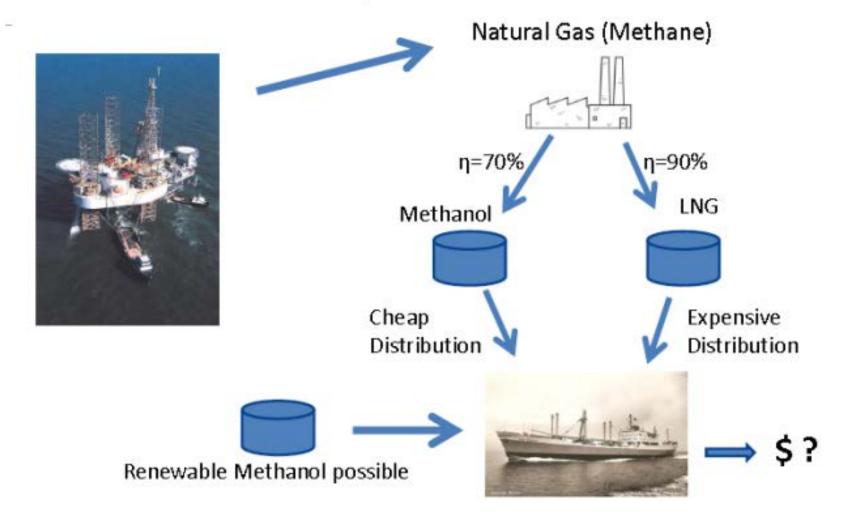
- Regulations driving change
- Emission Control Areas (ECA) 2015 near term focus
- In 2020, IMO targeting all marine fuels globally to be 0.5% sulphur







Methanol and LNG, two sides of the same coin





LNG CONCERNS

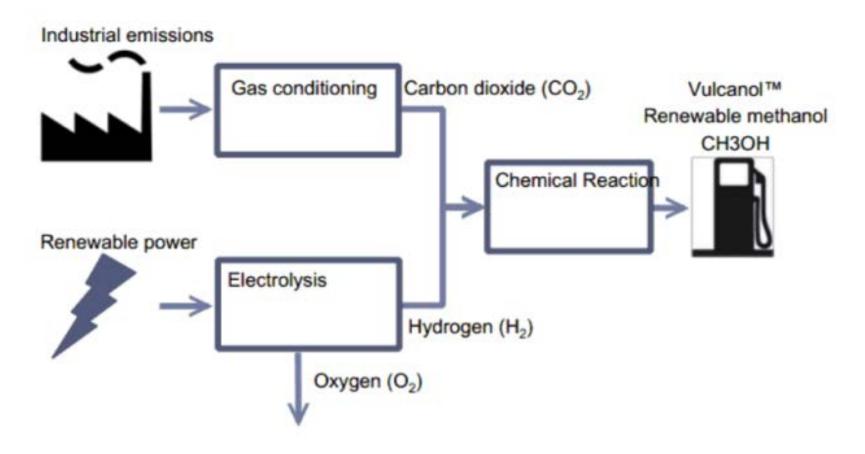
- Boil-off
- Pressure Increases
- Methane Slip
- Future regulations on fuel tank proximity to crew quarters

"George Olah CO₂ to Renewable Methanol Plant" Groundbreaking HS Orka Svartsengi Geothermal Power Plant, Iceland, October 17th 2009





Emissions to Liquids Production Process





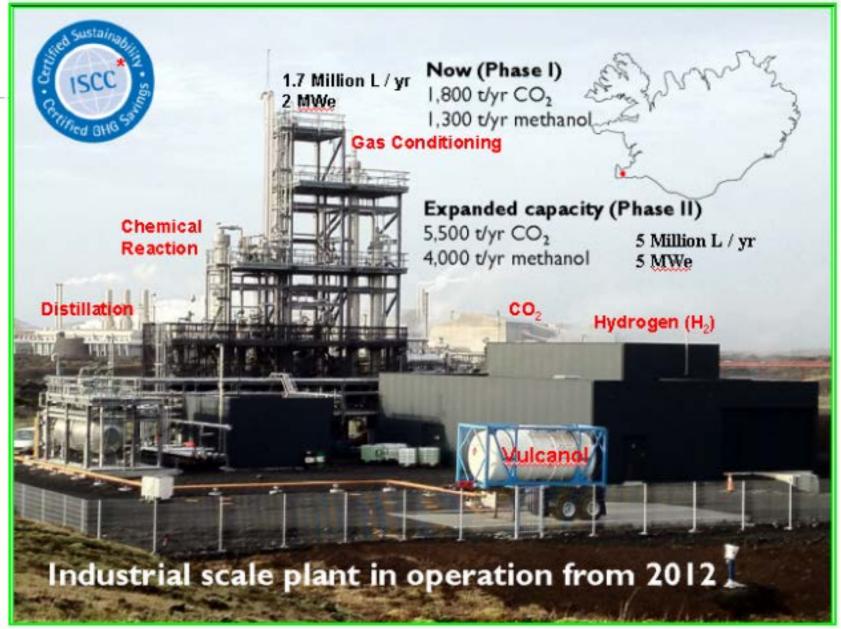






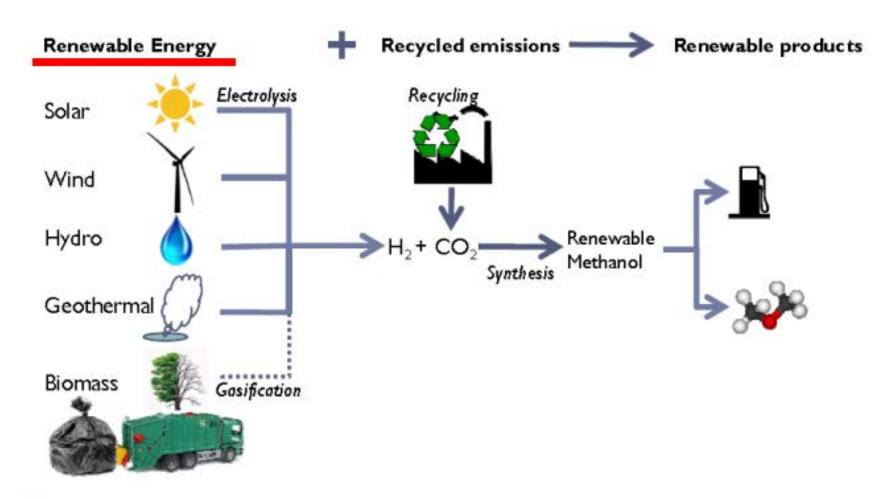








Diversity of Renewable Energy Sources:







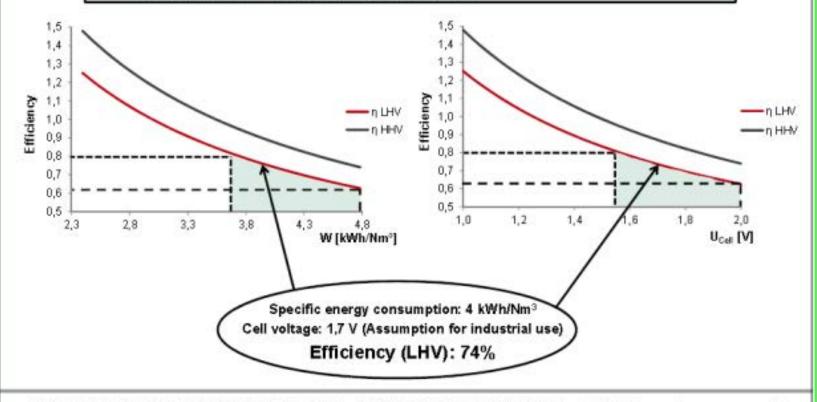
Efficiency of Alkaline Hydrogen Electrolysis



State of the art

Specific energy consumption (W): 3,7 - 4,8 kWh/Nm³ H₂

Efficiency: 63 – 80% (based on the LHV)

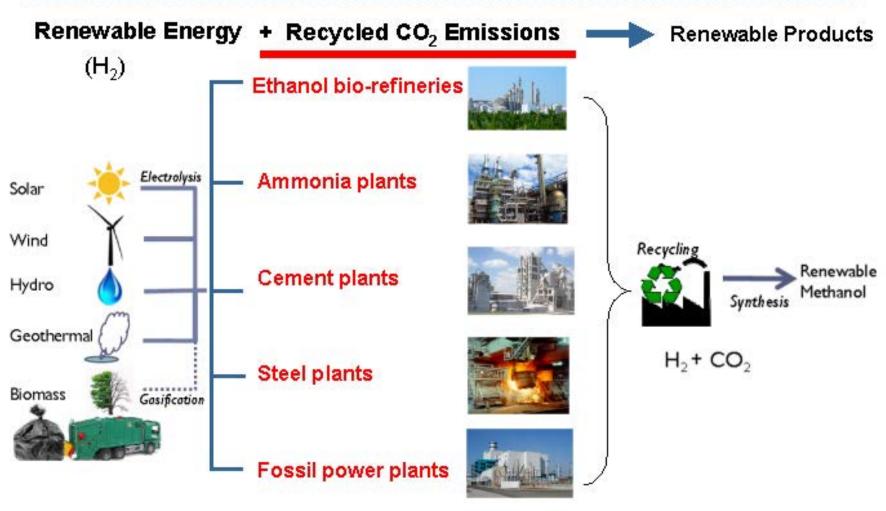


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And Diversity of CO₂ Sources

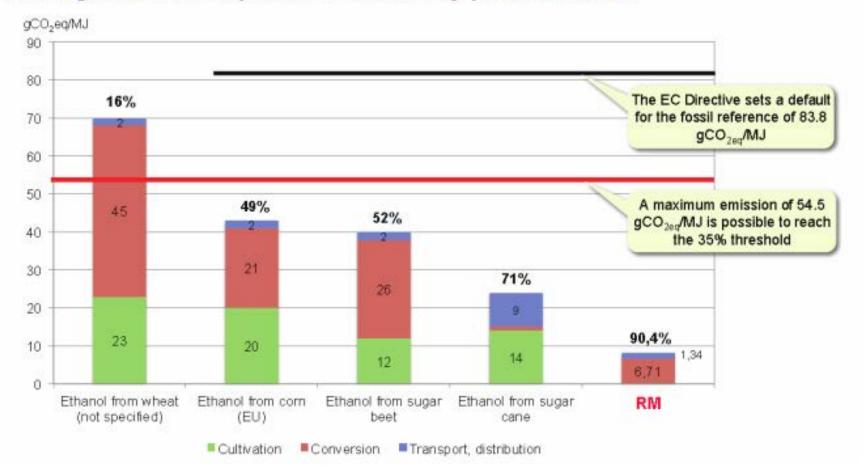








The Renewable Methanol achieves GHG savings of 90,4 %. Compared to existing alternative options this is a very positive result*



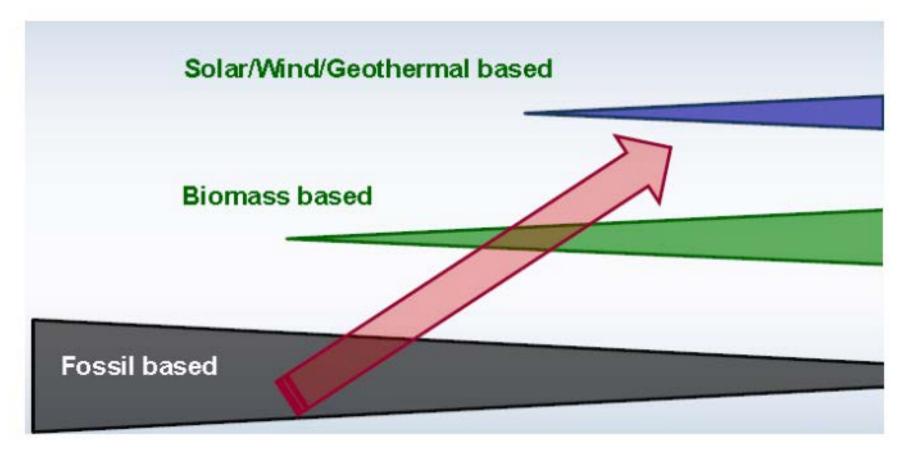
^{*} Based on biofuels default factors from RED



- First in building an industrial plant to commercialize power to liquid fuel
- First in utilizing kilotons of per year of CO₂ to make liquid fuel
- First in operating multi-megawatt electrolyzers for power to liquid fuel
- First in producing kilotons per year of renewable methanol for gasoline blending and biofuel manufacturing
- First in holding an International Sustainability and Carbon Certification for a renewable fuel with 90% reduction of CO₂ compared to gasoline



THE TRANSITION TO RENEWABLE METHANOL:



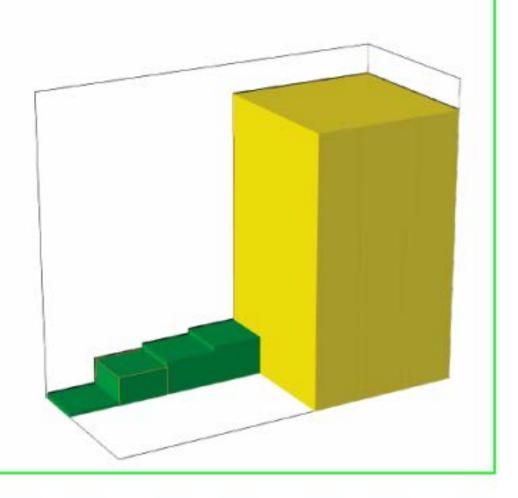


Limitless Potential for Renewable Power to Methanol Fuel

Power Consumption
Today: 17TW

Potential TW

Hydroelectric I
Biomass 7
Geothermal 6
Wind 8
Solar 488





CURRENT METHANOL USES

 Methanol is being used in an increasing number of energy applications

Traditional Uses (60% of Demand)

Formaldehyde

Wood Industry, Pharmaceuticals, Automotive



Dimethyl Terephthalate

Recyclable plastic bottles

Acetic Acid

Fleece, Adhesives, Paints



Methyl Chloride

Silicones



Energy & MTO (40% of Demand; High Growth)

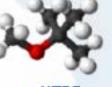


Fuel Blending



DME (di-methyl-ether)







MTBE



Marine Fuels



Methanol Characteristics

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	<u>Fuel</u>	Methanol	Natural Gas	Diesel	
	Density (kg/l)*	0.79	0,44 (as LNG)	0,85	
	Boiling point (°C)	65	-162	150-370	
	Flash point (°C)	11	-188	min. 60	
	Auto ignition (°C)	464	540	240	
-	Viscosity cSt at 20°C	~ 0,6	na	~ 13,5	
- 5	Octane RON/MON	109/89	120/120	526	
-	Cetane No.	3	-	45-55	
-	► LHV (MJ/kg)	20	50*	42	
	Flammability Limits, Vol%	7-36	5-15	1-6	
	Flame Speed (cm/s)	52	37	37	
	Heat of Evaporation (kJ/kg)	1178	na	233	
	Stoichiometric Air-Fuel Ratio	6,45	17,2	14,7	
	Adiabatic flame temp. (°C)	1910	1950	2100	



The DRAFT INTERNATIONAL CODE OF SAFETY FOR SHIPS USING GASES OR OTHER LOW FLASH POINT FUELS (IGF CODE). Under development. Intended fuels:

- Natural Gas (LNG) CH4 IGF Code Part A-1
- Propane C3H8
- Butane (i and n) C4H10
- Propane/Butane mixtures
- Ethyl alcohol C2H5OH IGF Code Part A-2
- Methyl alcohol CH3OH IGF Code Part A-2
- Hydrogen H2
- Dimethyl-ether CH3OCH3



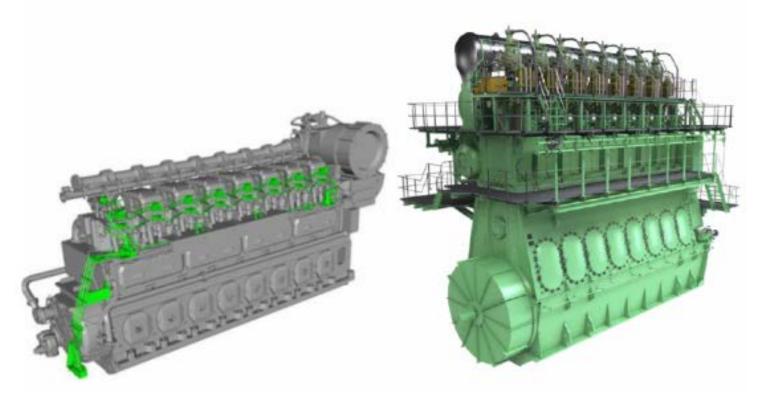
METHANOL MARINE DEMONSTRATIONS:

- Methanex tankers
 - MAN engines
- Spireth (methanol + DME)
 - MAN + Wartsila
- Stena Germanica (methanol conversion)
 - Wartsila
- Wartsila methanol / diesel
 - Engine testing + development



Methanol Marine Engine OEMs:

WÄRTSILÄ & MAN





Methanex: 7 Tankers to be Powered by Methanol





MAN S50ME-B9.3-LGI
Two stroke high pressure gas (methanol) engine

7 x 50 000 dwt Chemical Tankers on order.



Powered by MeOH





- SPIRETH Project technology confirmation
- Engine manufacturers developing methanol compatible engines
 - MAN 2 stroke engines
 - Wartsila 4 stroke engines
- Standards and regulations under development
 - Risk Classification Societies DNV, Lloyd's Register
 - Marine fuels regulations being updated for methanol





SPIRETH

- Bunker methanol CH₃OH
- Upgrade methanol to DME (CH₃OCH₃) onboard
- Use the DME in the existing diesel engines (only minor modifications required)



SPIRETH

- The benefits
 - Methanol/DME is available everywhere and limited only by the amount of feedstock (gas, coal, biomass etc)
 - The methanol market like gasoline is more transparent than LNG market
 - Methanol can be stored in hull tanks
 - Methanol/DME energy cost lower than for Marine Gas
 Oil
 - Methanol-DME operation will fulfil tier 3 NOx requirements without after treatment
 - Cost for conversion and/or adaption to methanol is a fraction of the comparable cost for LNG



Summary - Stena Germanica Methanol Conversion

- □ Complementary retrofit solution to reach Sulphur level of 0,1%
- □ Based on adapted known technology
- Minor modifications on engine (Common Rail addition)
- □ Dual fuel capacity, full redundancy Methanol / Diesel
- □ No reduction in efficiency running on methanol



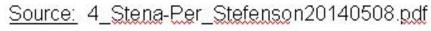






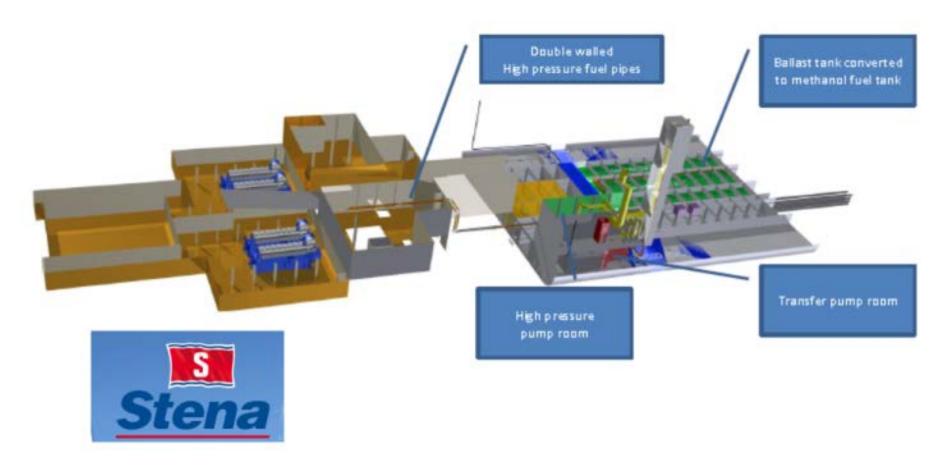








METHANOL RELATED MODIFICATIONS: STENA GERMINICA



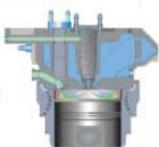


WARTSILA METHANOL ENGINE OPTIONS:

Use of Methanol in Internal Combustion Engines

Diesel Engine Concept for OBATE-DME *

The traditional HFO, or LFO, is replaced with OBATE-DME in a diesel engine. Since OBATE-DME is a gas at atmospheric pressure you need to pressurize the fuel supplyand return system. The challenge is the major modifications of the fuel injection system to handle fuel characteristics. As low viscosity, low heat value, low cetane number and high compressibility. Full output achievable.



Duel Fuel concept for Methanol

In this concept you replace the gas valve on a DF-engine, or complete the engine, with a methanol injector. You ignite the compressed premixed methanol-air mixture with a small pilot fuel diesel spray when the piston is close to TDC. Performance corresponding to the DF concept.



Methanol-Diesel concept for Methanol

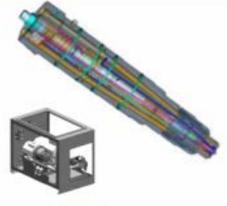
You inject the methanol at high pressure close to TDC and ignite with pilot diesel. By using the diesel principle you avoid knocking problems, which you might face using the methanol in a Duel Fuel engine. Area for adaption is the fuel injection system for the methanol. Full output achievable.



On-line conversion of MeOH to DME/MeOH/water mix



Wärtsilä Methanol-Diesel retrofit solution

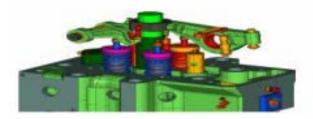


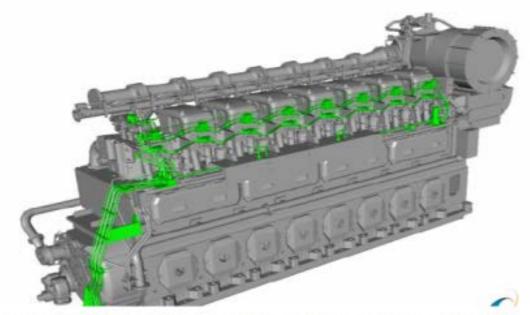
On-engine scope is limited to exchange of cylinder heads, fuel injectors and fuel plungers in existing fuel pumps.

A common rail system for methanol injection will be added on the engine.

In addition to the Engine related conversion includes the conversion kit a stand-alone high pressure methanol pump with belonging oil unit for supply of sealing oil and control oil to the fuel injectors.

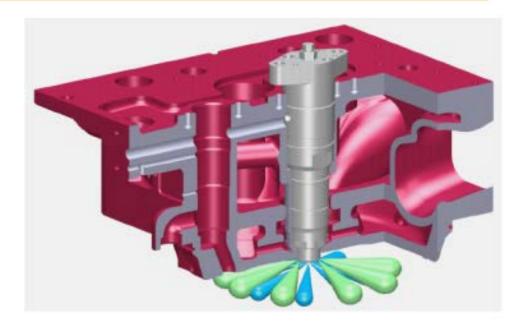








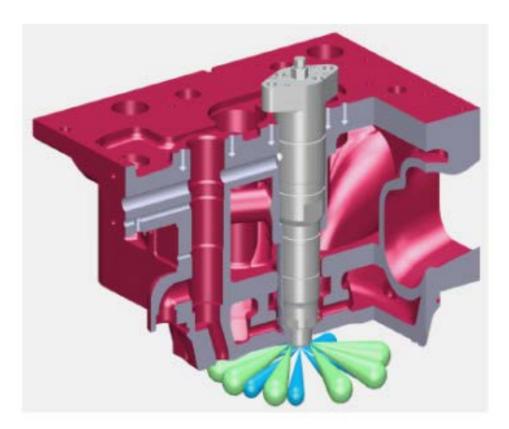
Wärtsilä Methanol-Diesel retrofit concept



Methanol is combusted according to the diesel process. The methanol is injected close to TDC and ignited by a small amount of pilot fuel - in this case traditional diesel fuel.



Methanol-Diesel Concept



- + no knocking (diesel combustion)
- + no power reduction
- + good load acceptance
- + cost effective adaption
- + low THC, CO and formal dehyde levels
- + good back-up fuel performance
- + HFO pilot fuel possible
- NOx higher than for DF concept
- depending on pilot fuel for ignition

* DF = Dual Fuel

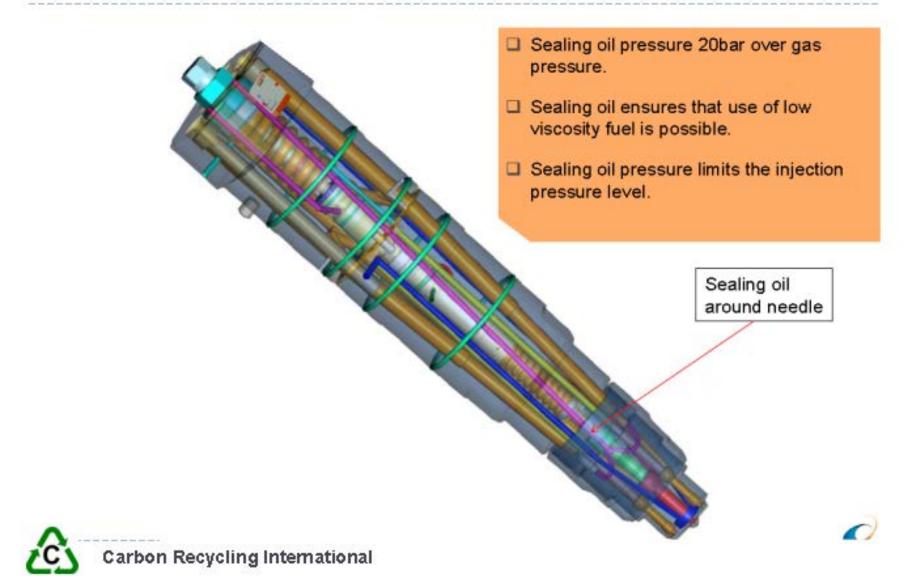


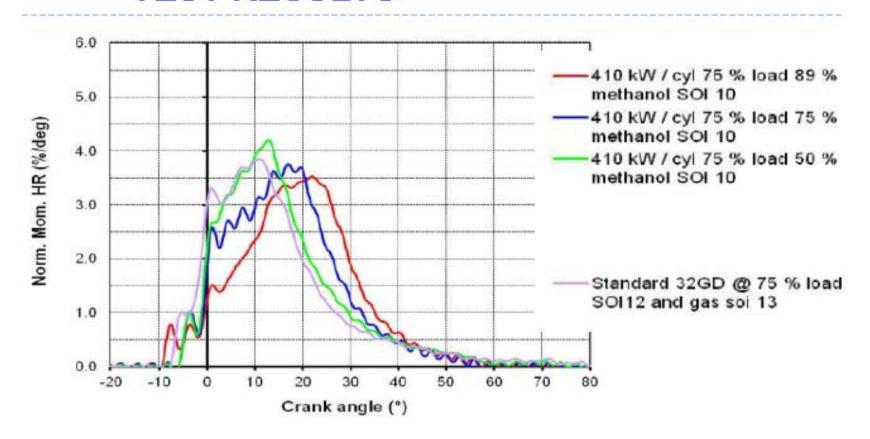
WARTSILA METHANOL- DIESEL TEST RESULTS

- NOx acceptable (Low Tier II values)
- CO acceptable (< 1 g/kWh)</p>
- THC acceptable (<1 g/kWh) and no "methane slip"</p>
- Very low PM (FSN ~ 0,1 with HFO as pilot)
- Formaldehyde emissions low 15-20 ppm (below TA-luft limit)
- ☐ Efficiency with methanol comparable to diesel
- No Formic acid detected in exhaust gases



FUEL INJECTION: The Heart of the Solution

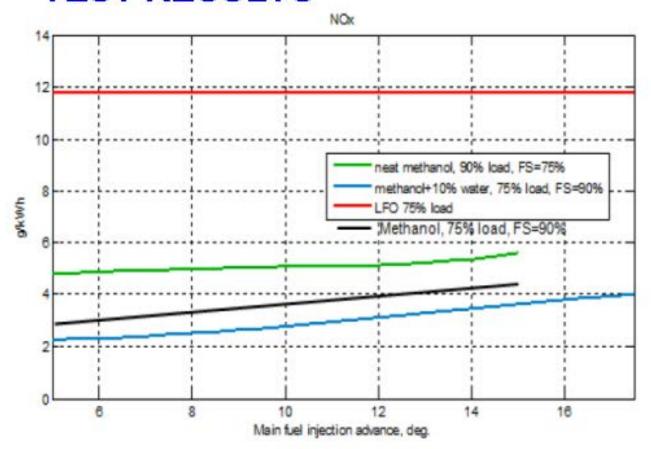




Optimal heat release



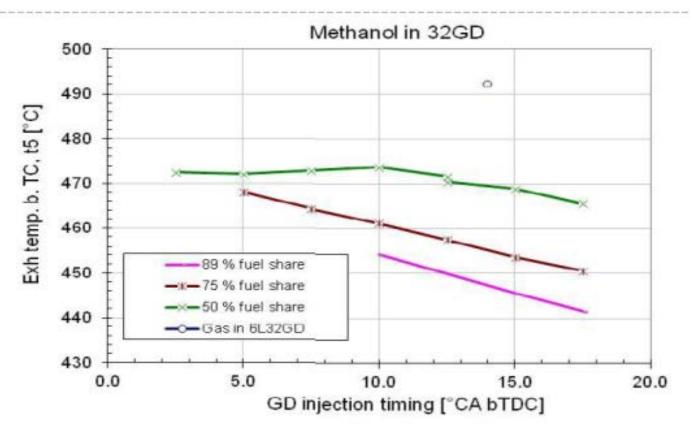
Source: Wartsila, Promsus Workshop, Gothenberg, Sweden - May 6, 2014



Reduced NOx (-70%)

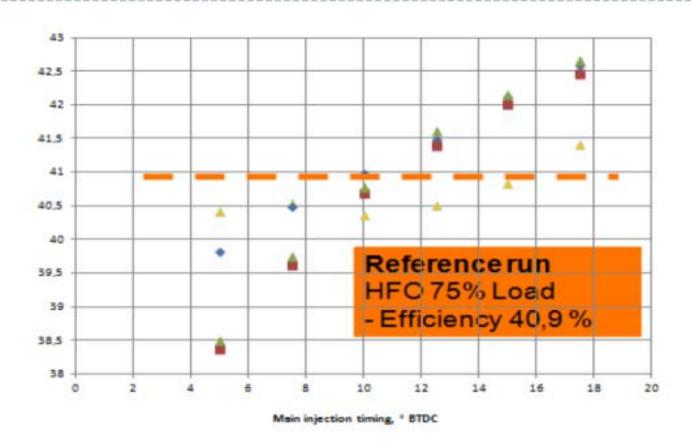


Source: Wartsila, Promsus Workshop, Gothenberg, Sweden - May 6, 2014



Lower Exhaust Temperatre (up to 50 C)



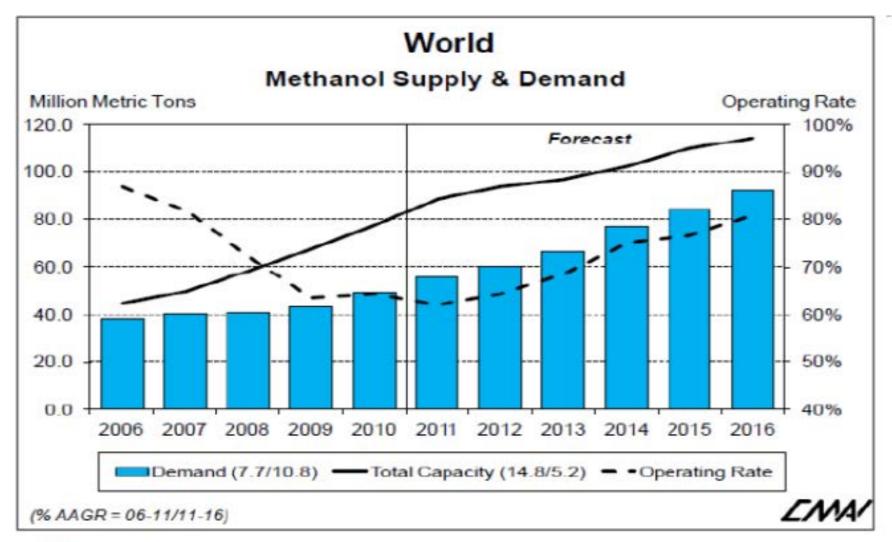


Same efficiency as diesel



Source: Wartsila, Promsus Workshop, Gothenberg, Sweden - May 6, 2014

PROMISING METHANOL SUPPLY OUTLOOK





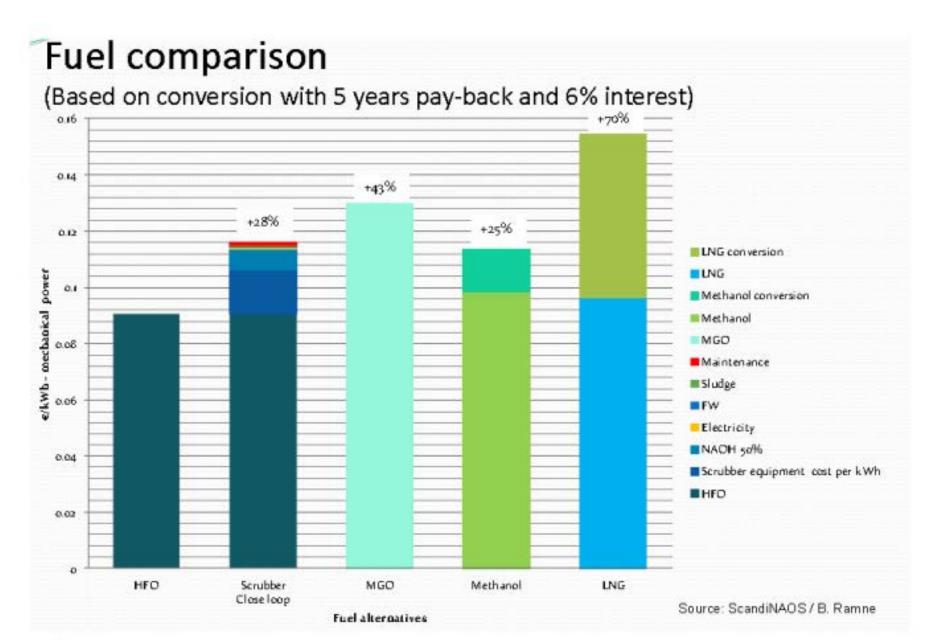
US Shale Gas: A Methanol Renaissance?

The shale gas powered resurgence of the methanol industry in the United States:

"If you add up all this activity we're talking about getting to the point where the U.S. will have some 16 million metric tons of production capacity in the next three years. To put that in perspective, our current chemical demand is about 6.5 million metric tons so you could have an overhang of 10 million metric tons (3.3 billion gallons) looking for a market."

Greg Dolan, Methanol Institute July, 2014







CONCLUSIONS:

- Methanol marine engines are maturing rapidly
- Methanol offers the lowest cost of compliance
- Sustainable methanol will continue to accelerate
- LNG infrastructure, in-use boil-off inefficiencies, and ship retrofit capital cost concerns pose far greater challenges than methanol fuels
- Methanol is available globally at reasonable and competitive prices
- The demonstration of methanol in EU marine applications will lead the way
- CA ports should incorporate methanol in their range of options for serious consideration



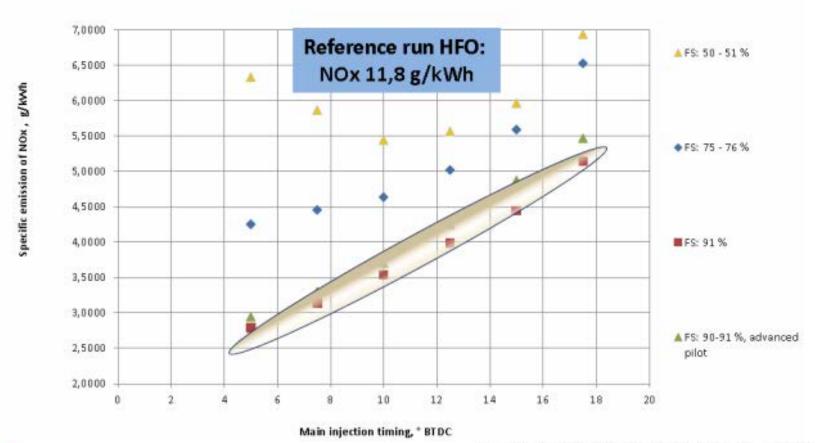




SUPPLEMENTAL SLIDES



Tests on Methanol - NOx (METHANOL-DIESEL)





SOURCE: Wartsila, March 21, 2013, Gothenberg Workshop

WARTSILA METHANOL-DIESEL TEST RESULTS:

Engine: 4L32LNGD

Output: 410kW/Cylinder

Compression ratio: 13.8:1

- □ NOx 3-5 g/kWh (Low Tier II)
- □ CO acceptable (< 1 g/kWh)</p>
- □ THC acceptable (< 1 g/kWh)</p>
- □ Very low PM (FSN ~ 0,1 with HFO as pilot)
- □ Formaldehyde emissions low ~ below TA-luft
- ☐ Efficiency comparable to running on diesel
- □ No Formic acid detected in exhaust gases





Source: Wartsila, Promsus Workshop, Gothenberg, Sweden - May 6, 2014

LNG Concerns:

The fact that LNG is stored at -260 degrees F and that its energy density is only 60 percent that of diesel fuel means there must be significant changes in a ship's engines, its tankage and fuel handing systems, and its overall design and operation.

1) BOIL-OFF: With cyrogenic fuels like LNG, there is a possibility for de – bunkering (or emptying the fuel tanks). This step is necessary when a ship is to be anchored for an extended period of time. Unless special LNG de – bunkering facilities are available in the port, the gas would boil off, causing huge methane losses to the atmosphere. In case of grounding accidents, a technique for de – bunkering would also be necessary.

LNG Concerns (cont.)

- 2) PRESSURE INCREASES: such increases are possible when consumption occurs below the natural boil-off rate, which will happen if there is no re-liquefaction plant available onboard. Re-liquefaction of boil-off gas requires about 0.8 kWh/kg gas. One large LNG carrier, such as Qatar Q-max, requires 5–6 MW of re-liquefaction power, corresponding to a boil-off rate of 8 tons/hour.
- 3) METHANE SLIP: inadvertent CH4 emissions from larger marine engines burning NG can occur. Methane slip will be present, especially on four-stroke, dual-fuel NG engines, partly from the scavenging process in the cylinder and partly from the ventilation from the crank case, which is being led to the atmosphere.
- 4) FUTURE RETROFIT REGULATIONS: prohibitions may be placed on LNG tanks situated directly below accommodation which could cause difficulties in retrofitting certain ships.

Methanol may be permitted as a fuel onboard cargo ships

SOLAS II-2, Part B, Regulation 4.2.1 states...

- 4.2.1.1 except as otherwise permitted by this paragraph, no oil fuel with a flashpoint of less than 60° C shall be used;
- 4.2.1.4 in cargo ships the use of fuel having a lower flashpoint than otherwise specified in paragraph 2.1, for example crude oil, may be permitted provided that such fuel is not stored in any machinery space and subject to the approval by the Administration of the complete installation.



No Rules or Regulations specifically for use of methanol as a fuel onboard ships, but...

Rules for carriage of methanol

- International Code for the Construction and Equipment of Ships Carrying Dangerous Chemicals in Bulk (IBC Code) - IMO
- Rules and Regulations for the Construction and Classification of Ships for the Carriage of Liquid Chemicals in Bulk - Lloyd's Register

Rules for use of low flash point fuels

- International Code for the Construction and Equipment of Ships Carrying Liquefied Gases in Bulk (IGC Code) - IMO
- Rules and Regulations for the Construction and Classification of Ships for the Carriage of Liquefied Gases in Bulk - Lloyd's Register





EPA METHANOL ASSESSMENT:

- Soil Biodegradation is the major route of removal of methanol from soils. Several species of Methylobacterium and Methylomonas isolated from soils are capable of utilizing methanol as a sole carbon source (CHEMFATE 1994).
- Water Most methanol is removed from water by biodegradation. The degradation products of methane and carbon dioxide were detected from aqueous cultures of mixed bacteria isolated from sewage sludge (CHEMFATE 1994). Aerobic, Gram-negative bacteria (65 strains) isolated from seawater, sand, mud, and weeds of marine origin utilized methanol as a sole carbon source (CHEMFATE 1994). Aquatic hydrolysis, oxidation, and photolysis are not significant fate processes for methanol (HSDB 1994).
- Biota Bioaccumulation of methanol in aquatic organisms is not expected to be significant based on an estimated bioconcentration factor of 0.2 (HSDB 1994).

